Roll No.

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### 2E1026

# B.Tech. I Year II Semester (Back) Examination-2014 Common to all branches of Engg. 206 Engineering Chemistry - II

Time: 3 Hours

Maximum Marks: 80

Min. Passing Marks: 24

#### Instructions to Candidates:

Attempt any five questions, selecting one question from each unit. All questions carry equal marks. (Schematic diagrams must be shown wherever necessary. Any data you feel missing suitably be assumed and stated clearly. Units of quantitics used/calculated must be stated clearly.)

#### Unit - I

- 1. a) Define cracking? What are the advantages of catalytic cracking. Describe with neat and labelled diagram of fixed bed catalytic cracking (12)
  - b) Write on "Characteristics of a good fuel".

(4)

#### OR

- 1. a) Explain the manufacturing of synthetic petrol by Bergius process. (8)
  - b) Write short notes on any two
    - i) Octane number
      - ii) Oil Gas
      - iii) Carbonization of coal

 $(2\times4)$ 

#### Unit - II

- 2. a) Define calorific value of fuel? Explain the determination of CV of gaseous fuel by Junker's calorimeter. (8)
  - b) Calculate the gross and net calorific value of a coal sample, having the following composition- C=85%, S=1%, H=5%, O=8% & rest is ash. (8)

		i) Flue gas analysis by Orsat's apparatus	
		ii) Significance of ultimate analysis	(4+4)
s. T.	<b>b</b> )	A sample of coal was found to contain the following % composition C H=5%, O=2%, N=3% & rest is ash. Calculate the minimum weight of c and air required for complete combustion of 1kg of coal.	
÷		Unit - III	ti e
3.	a)	What is phase rule? Discuss the term phase, component and degree of frowith example.	eedom (8)
18	b)	Explain sulphur system in detail.	(8)
	٠.	OR	9
3.	·a)	What is reduced phase rule? Explain phase diagram of Bi-Cd system	(8)
	b)	Write short note on any two	
e ·		i) Application of Ag - Pb system	
10	10	ii) Triple point	
	Ę.	iii) Eutectic point	(2×4)
7			
		Unit - IV	
4.	a)	Define optical fibres? Describe the construction, working and application optical fibers.	ions of (10)
	b)	Write notes on applications of conducting organic polymers	(6)
·	6	OR	
4.	a)	What are super conductors? Discuss their properties and uses	(10)
	b)	Write notes on fullerenes	(6)

## Unit - V

5.	a)	What is corrosion? Explain electro chemical corrosion/wet corrosion mechanism in detail (10)
	<b>b</b> )	Discuss concentration cell corrosion (6)
		OR
5.	a)	Write any four factors to influencing corrosion and any four methods to control the corrosion (10)
X)	b)	Distinguish between Galvanizing and Tinning. (6)